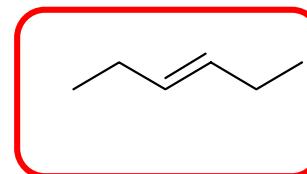
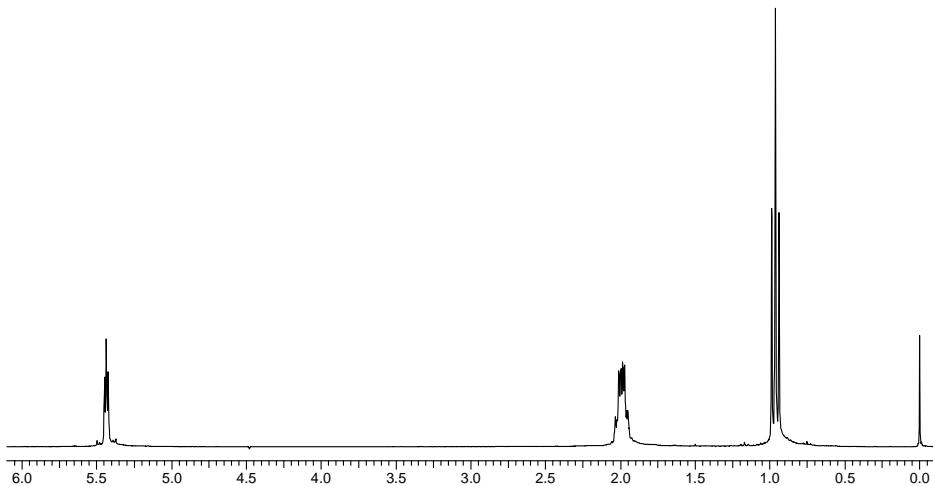
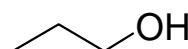
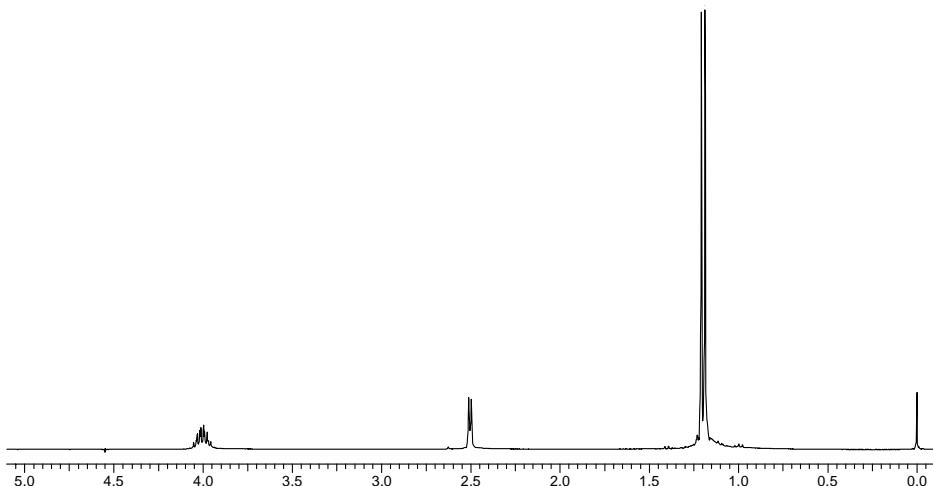
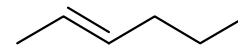


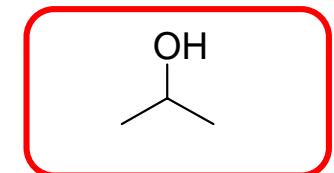
# Homework



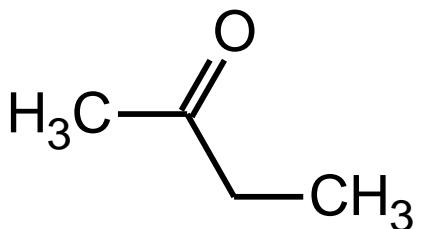
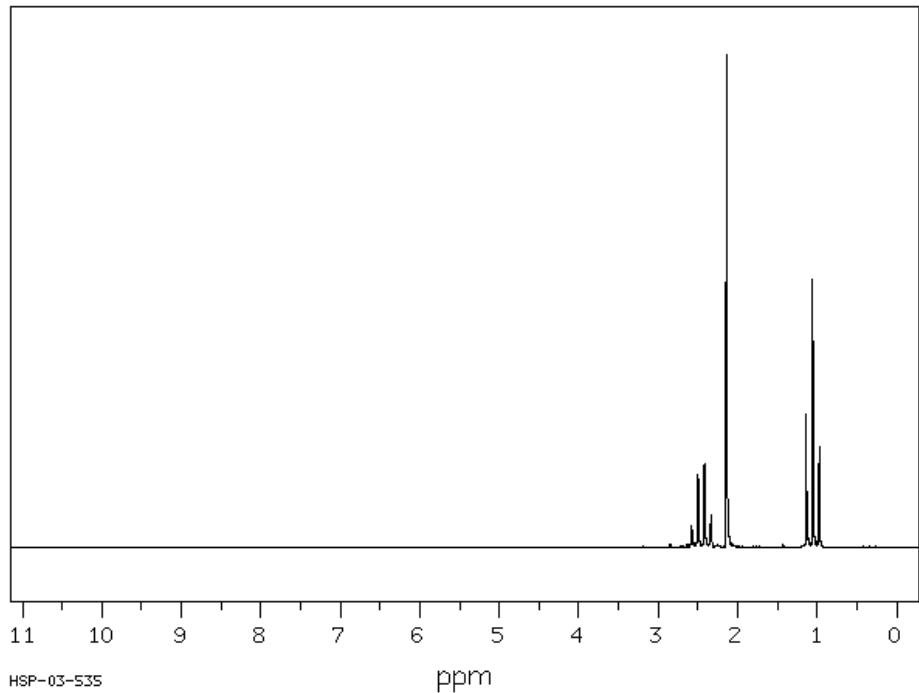
or



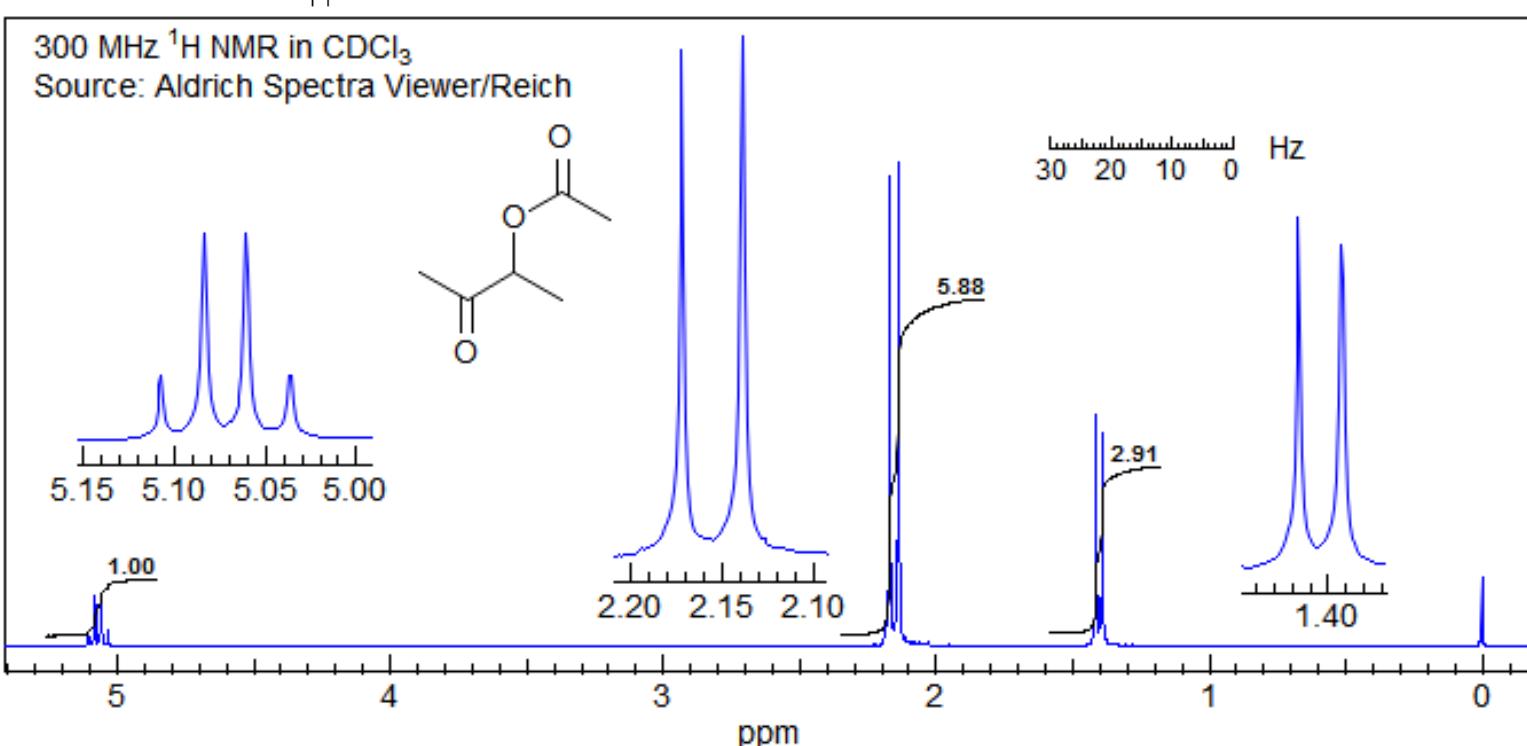
or

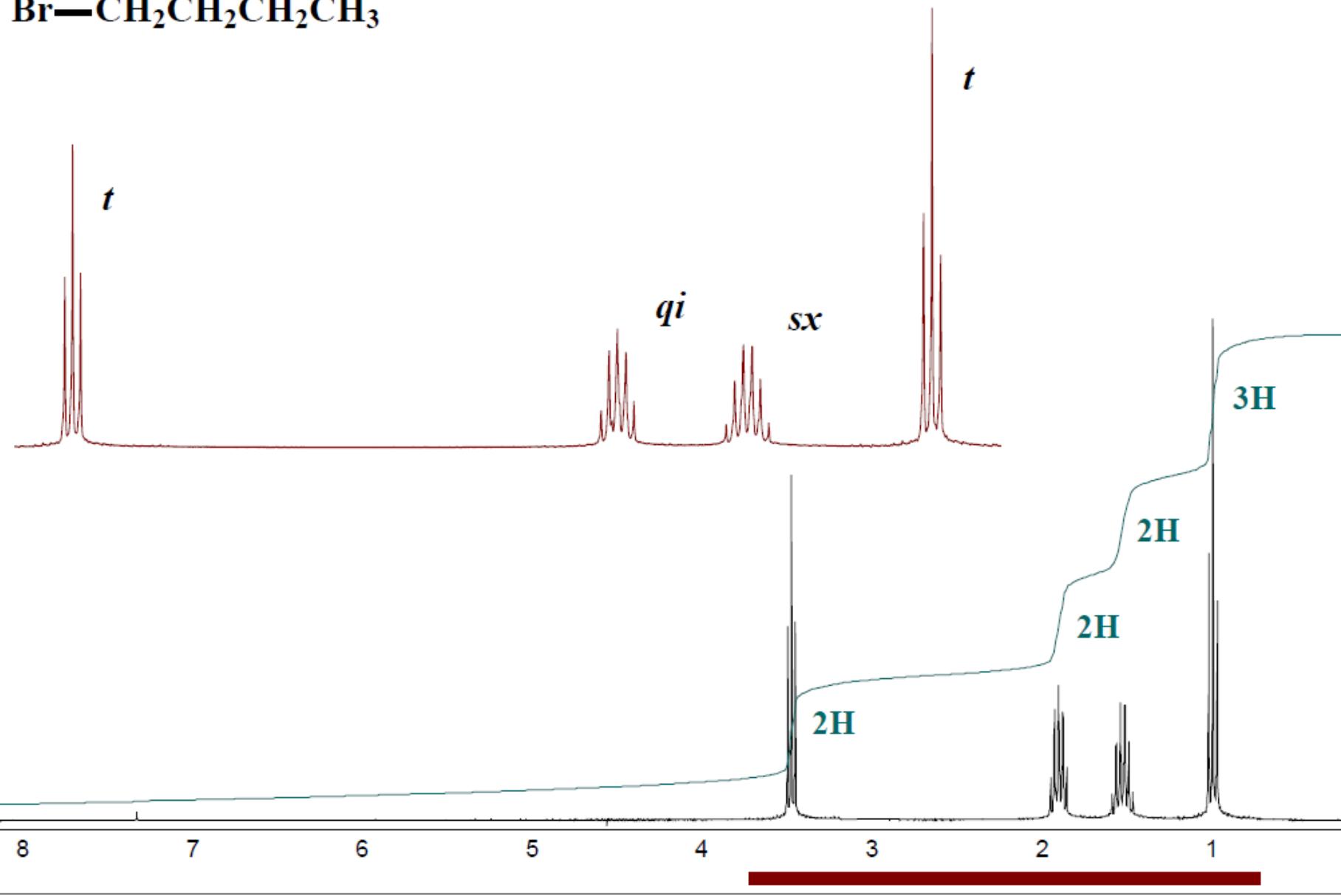


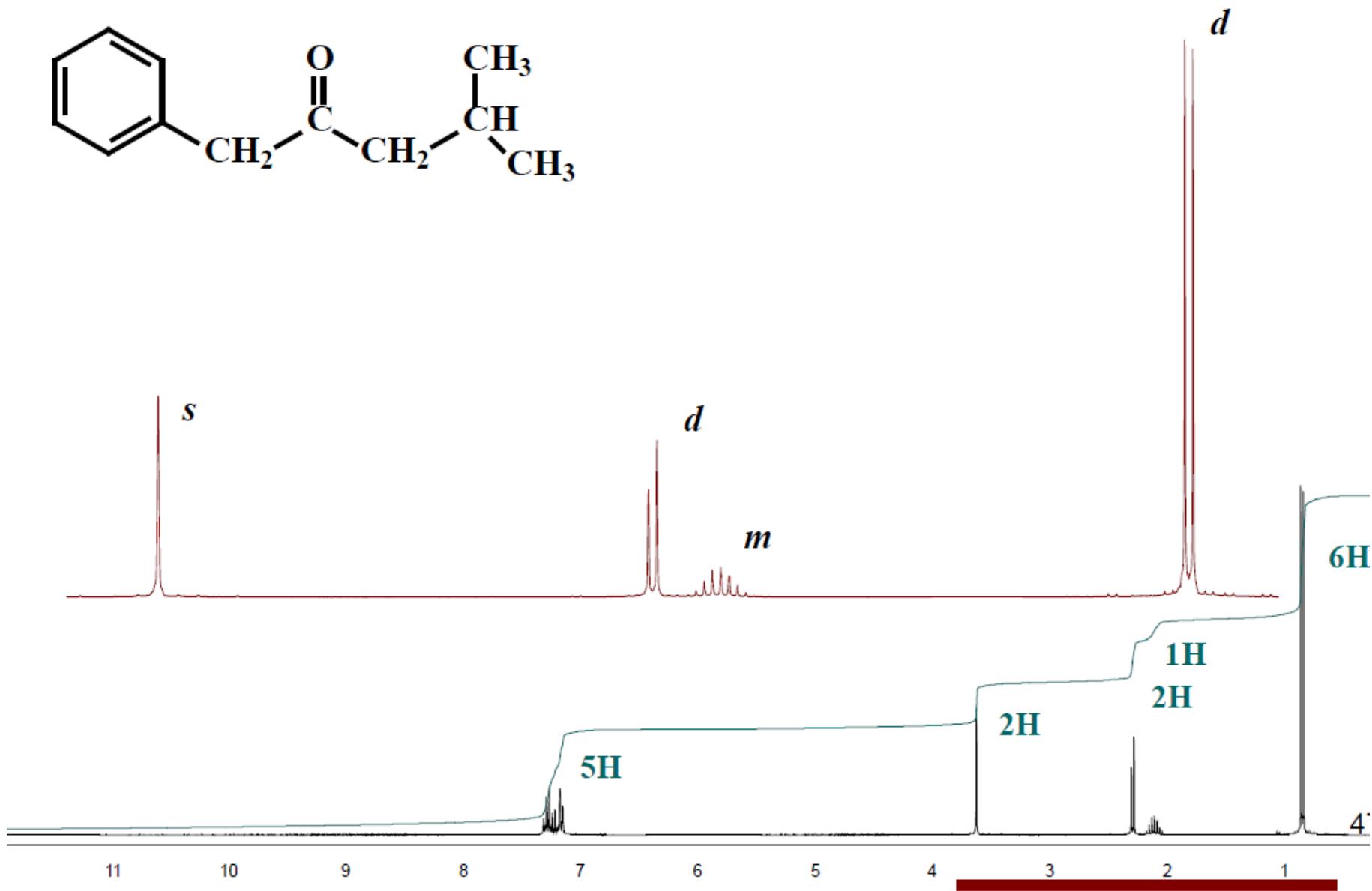
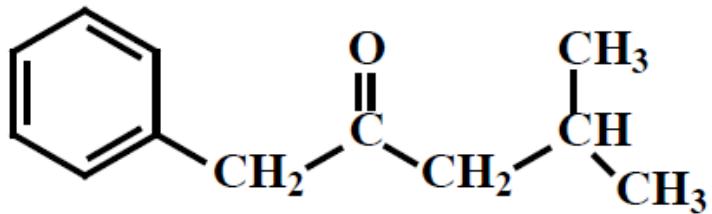
Coupling with OH is not always visible, depends on solvent

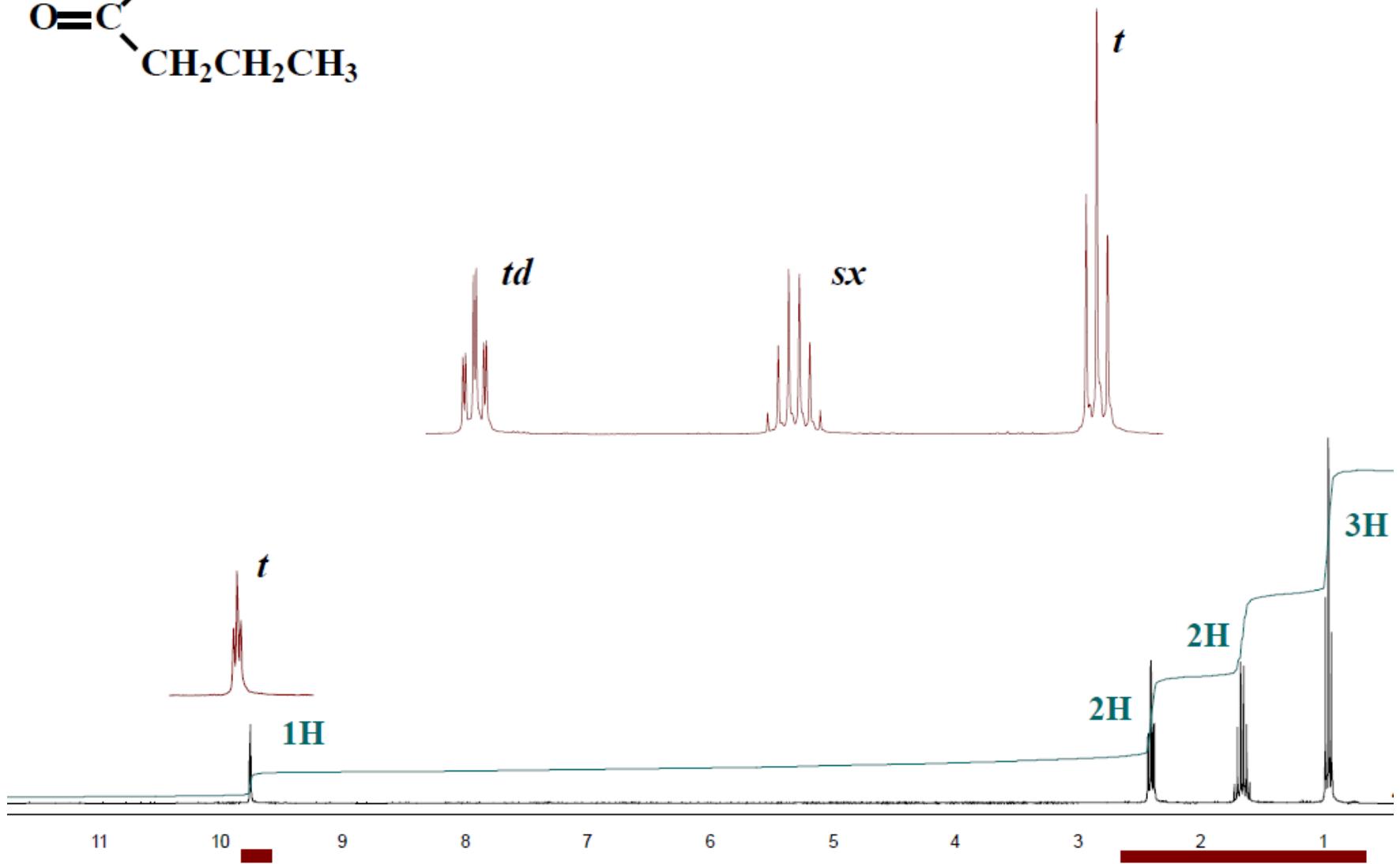
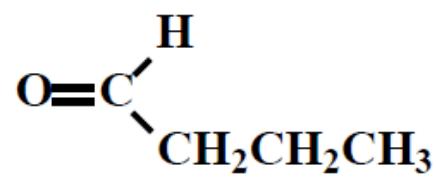


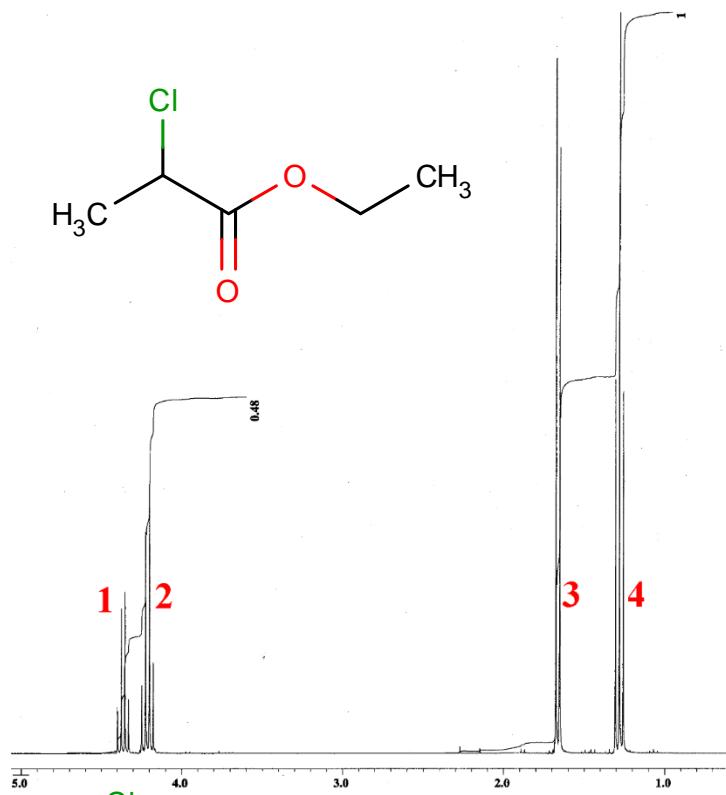
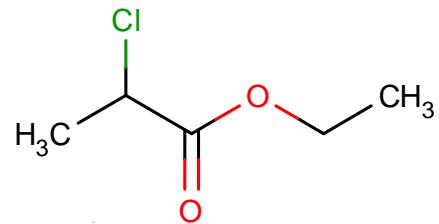
HSP-03-535



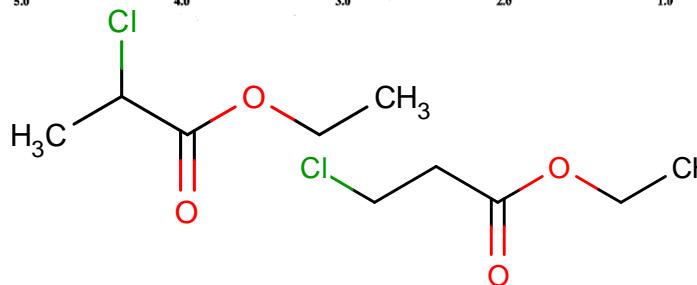
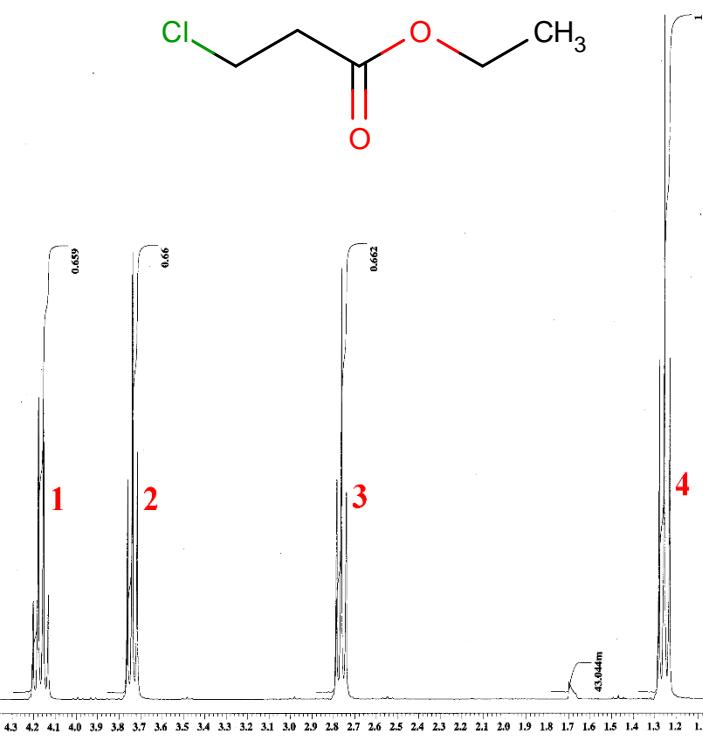
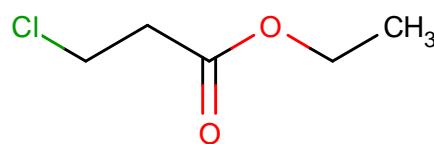


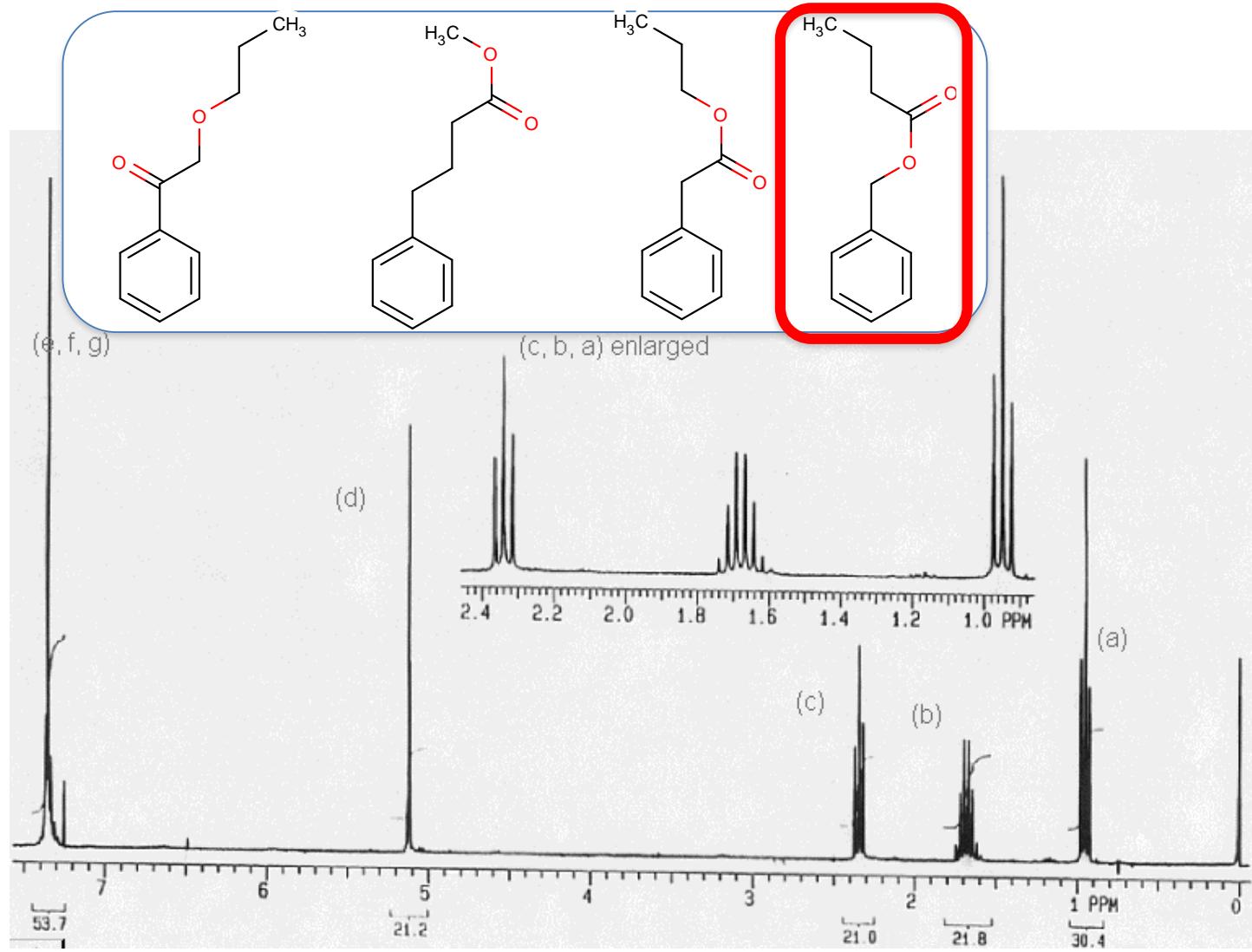






Ethyl chloropropionates, which is which?

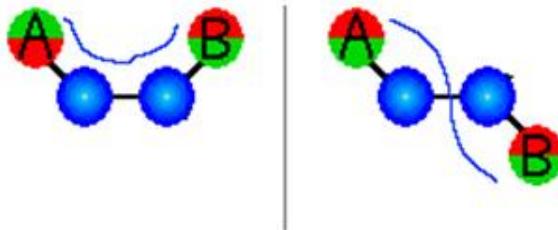




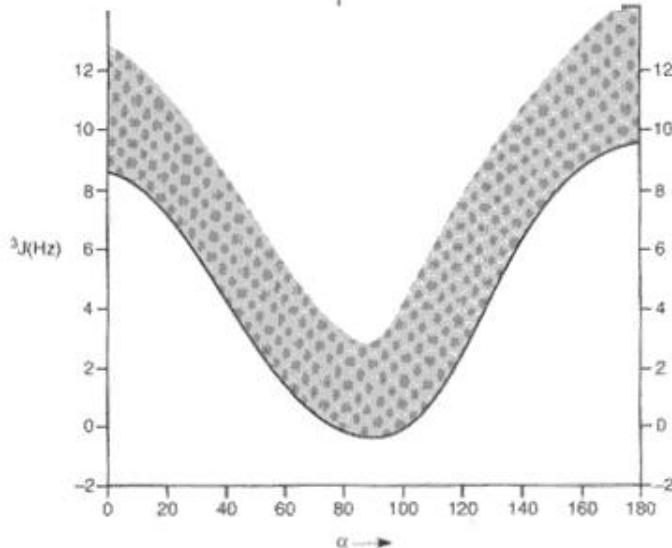
# Information about

1.Coupling partners → Constitution

2.Bond angle → 3D structure!!!



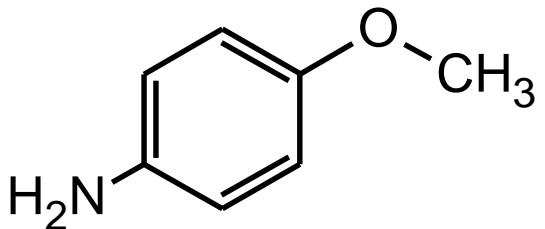
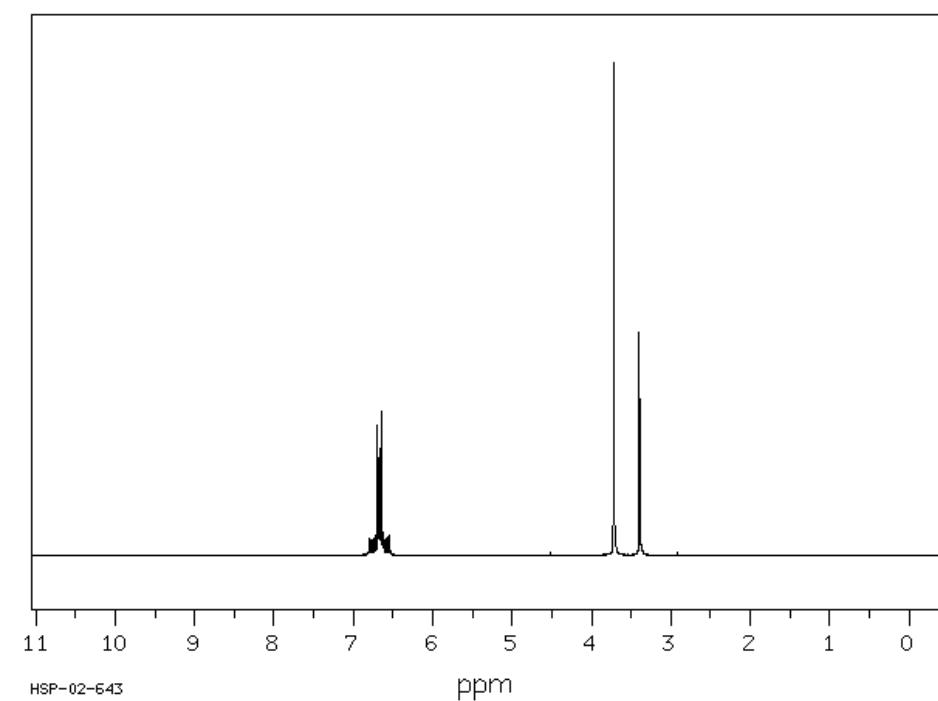
Karplus equation



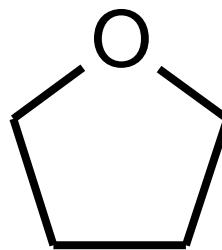
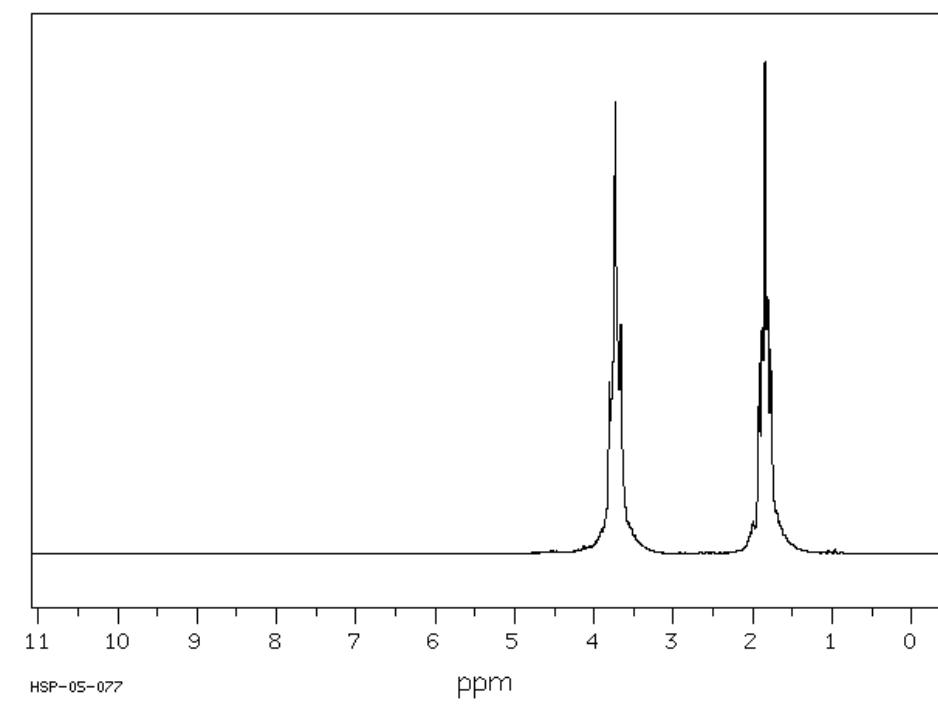
# Coupling constants

$^2J$ aliphatic	10-16	cis	12
$^2J$ sp <sup>2</sup>	1-2	trans	16
$^3J$ aliphatic free rotation	6-7	ortho	7-9
ax-ax	12	meta	2
ax-eq, eq-eq	2-3	para	1

Lots of experimentally collected values can be find in tables



Non first order spectra!!!



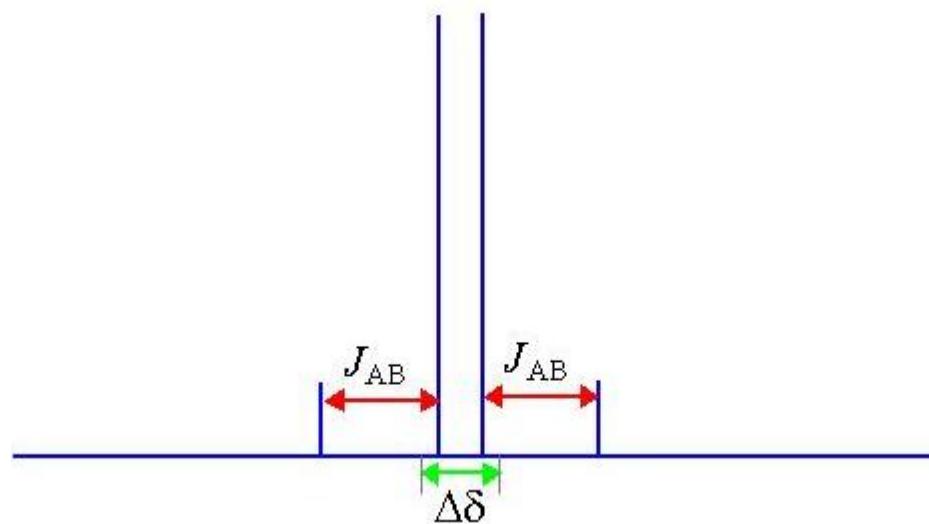
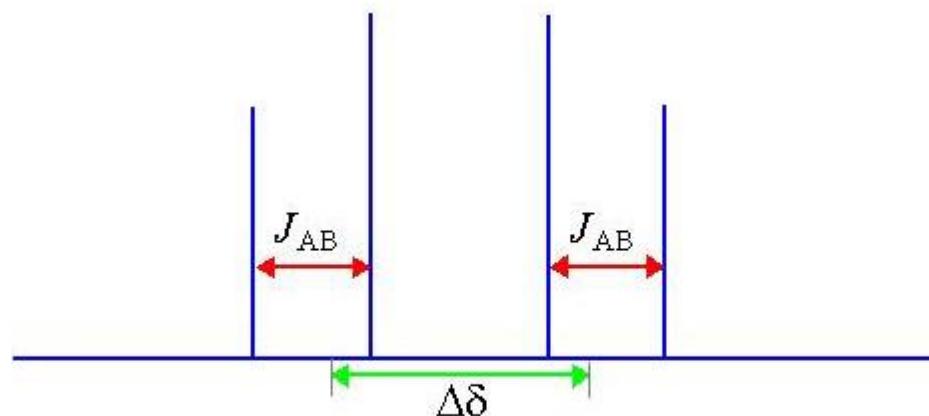
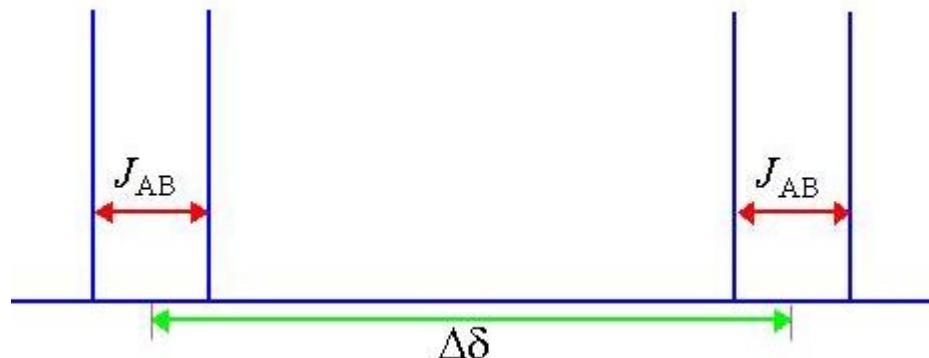
# Non first order spectra

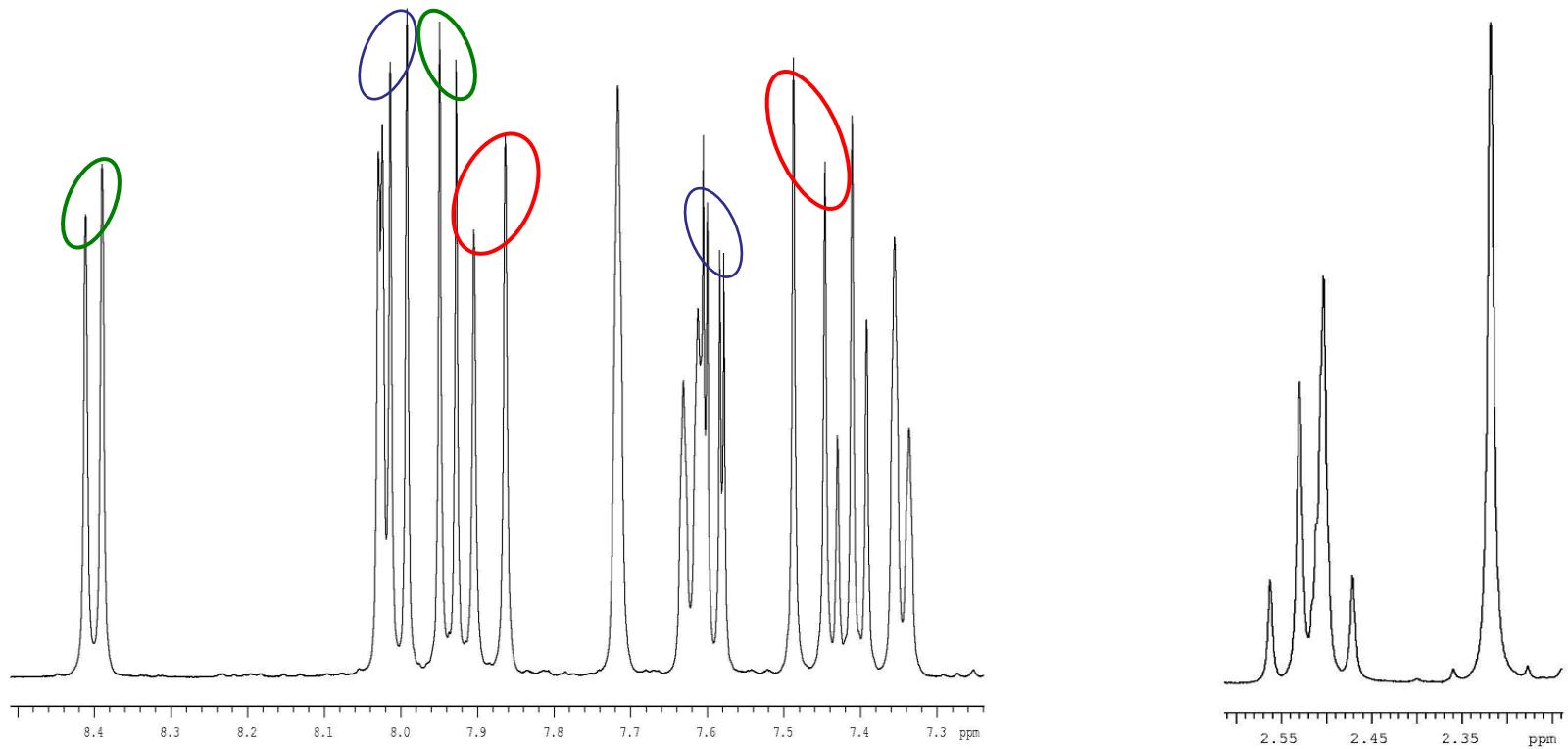
First order:  $10^*J < \Delta\omega$

$J / \Delta\omega$  increasing, the spectrum can change totally

Spin systems named by the letters of alphabet

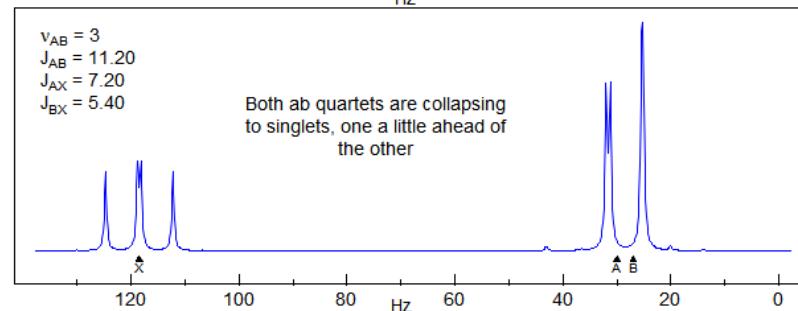
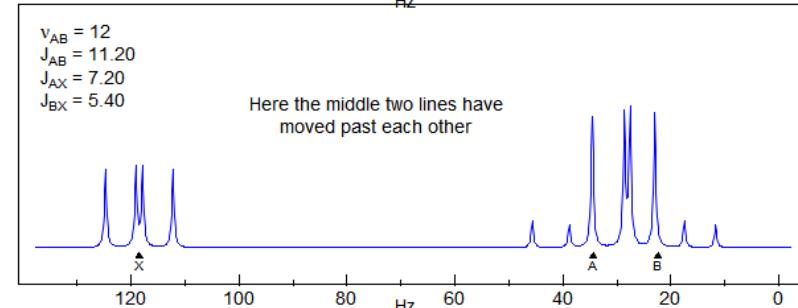
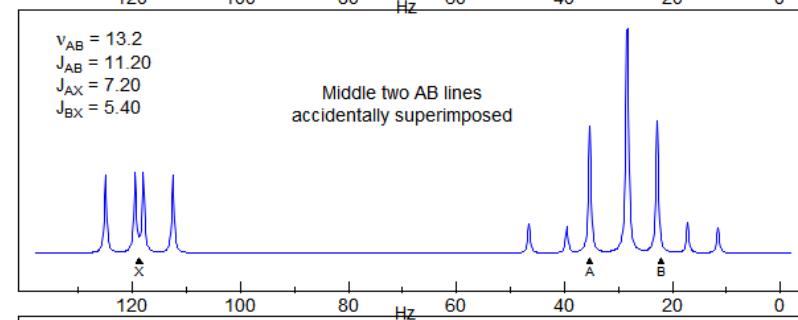
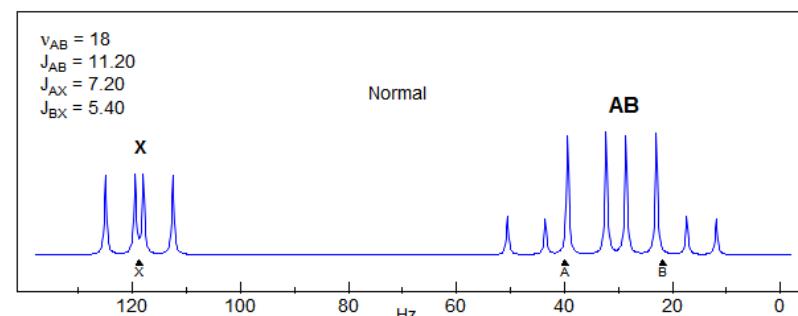
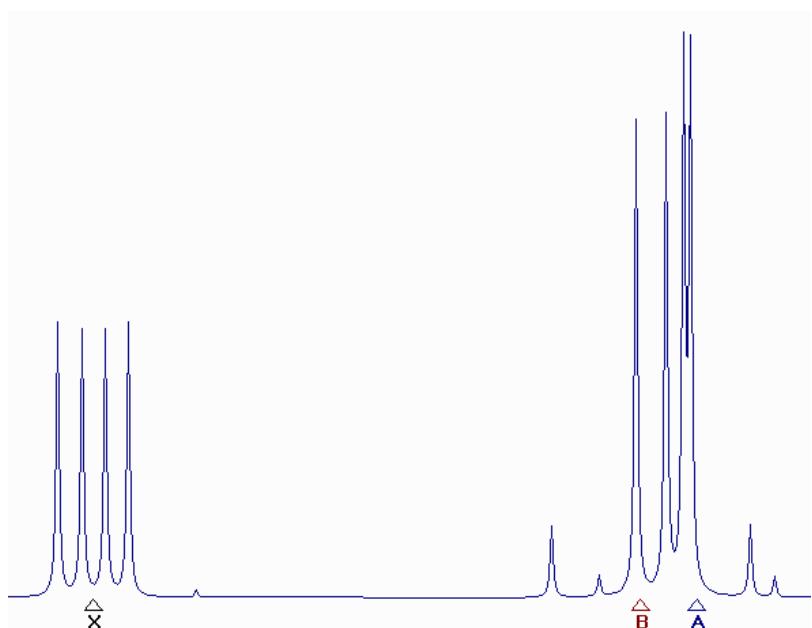
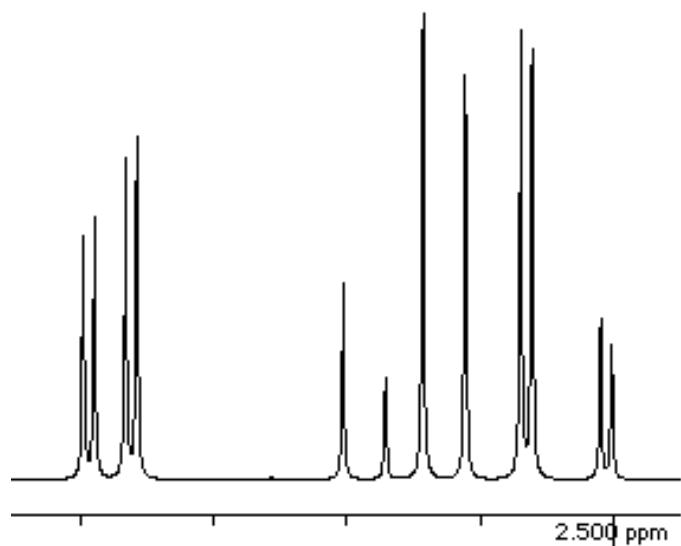
AB

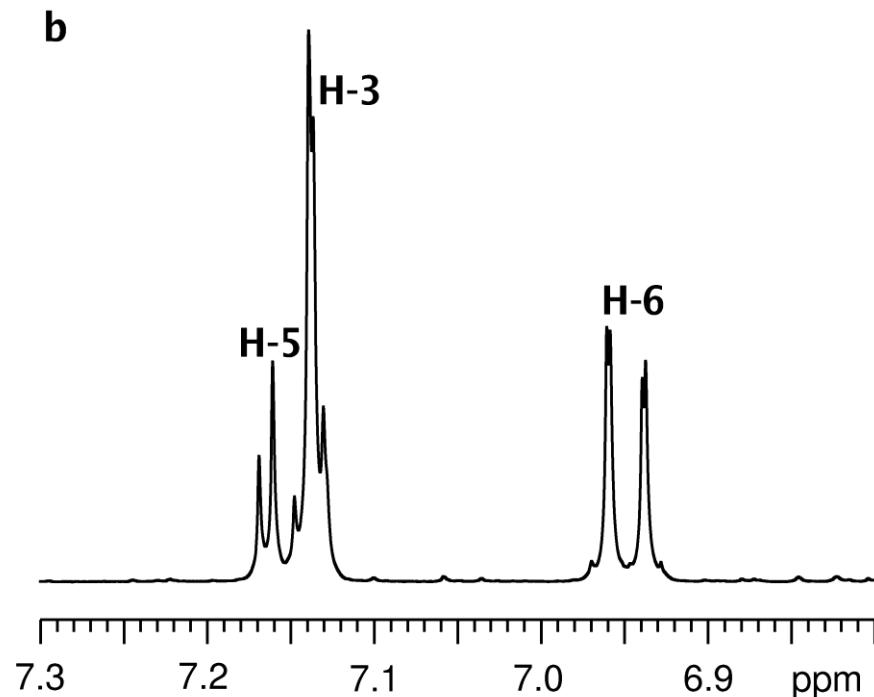
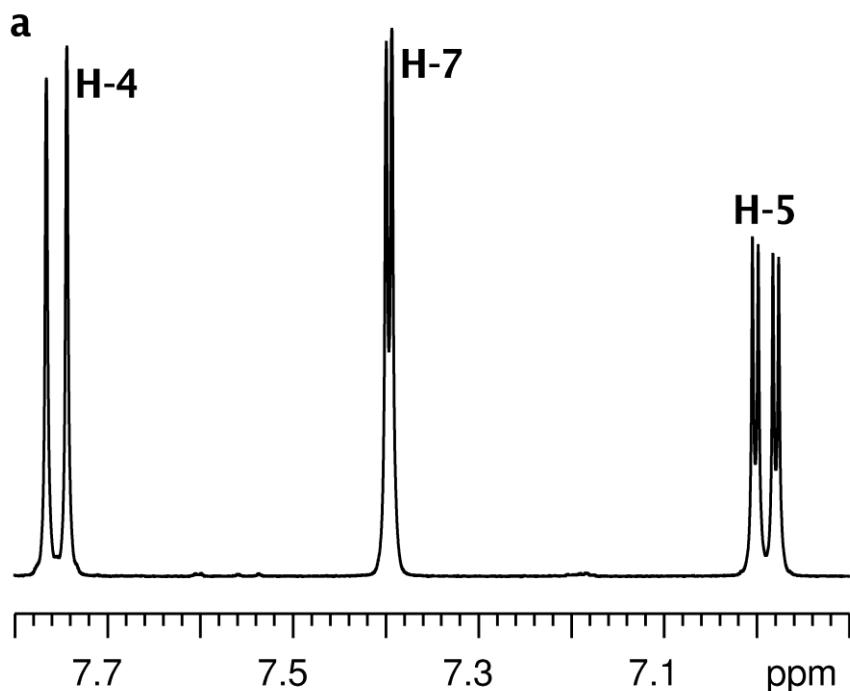
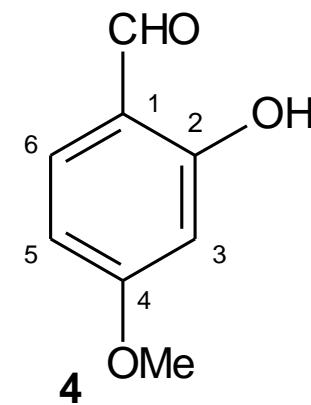
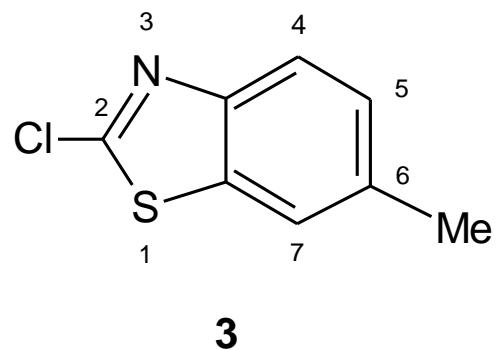




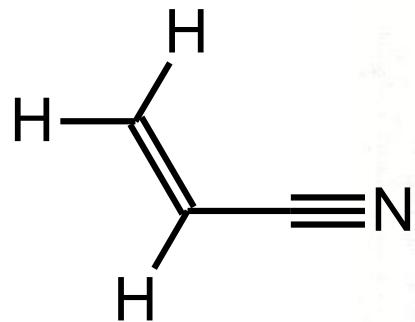
Mon Oct 4 12:58:48 1993: (untitled)  
W1: 1H Scale = 15.02 Hz/cm  $\Delta(1/2) = 0.50$  Hz

ABX

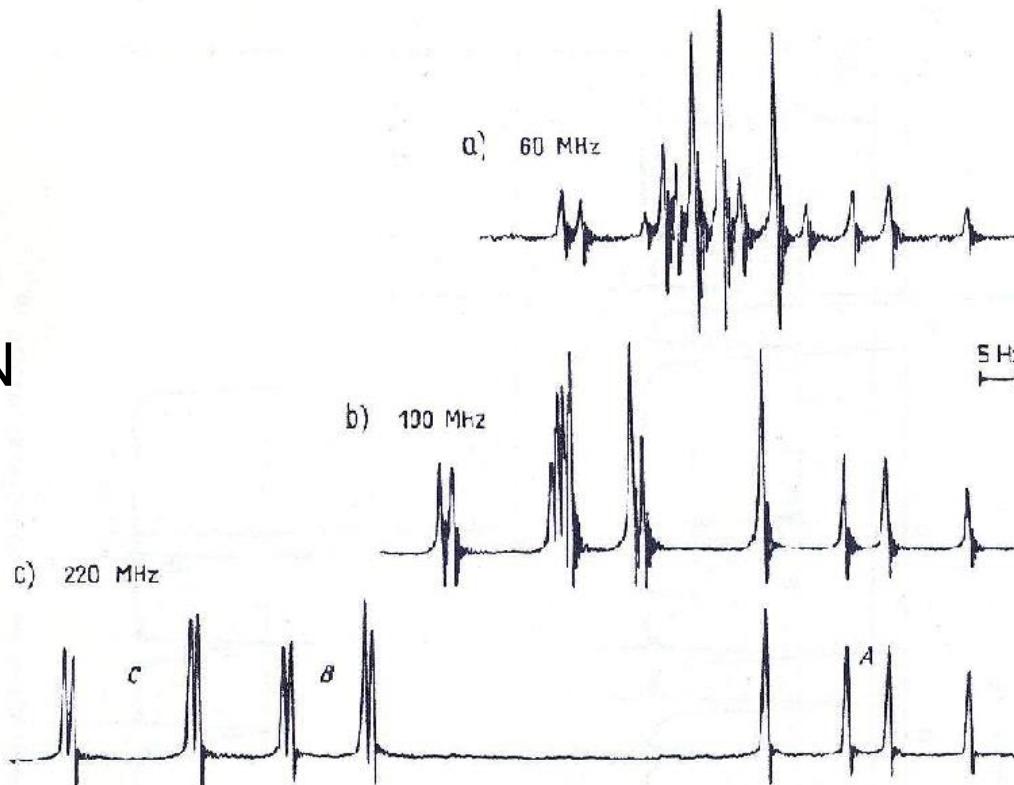




Acrylnitrile



400 MHz



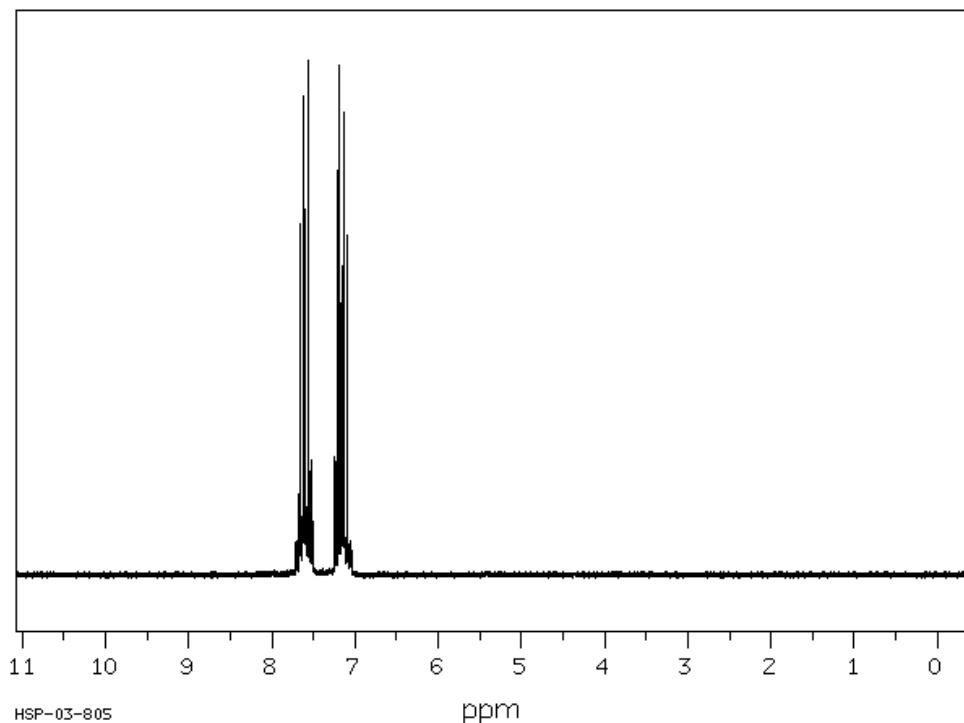
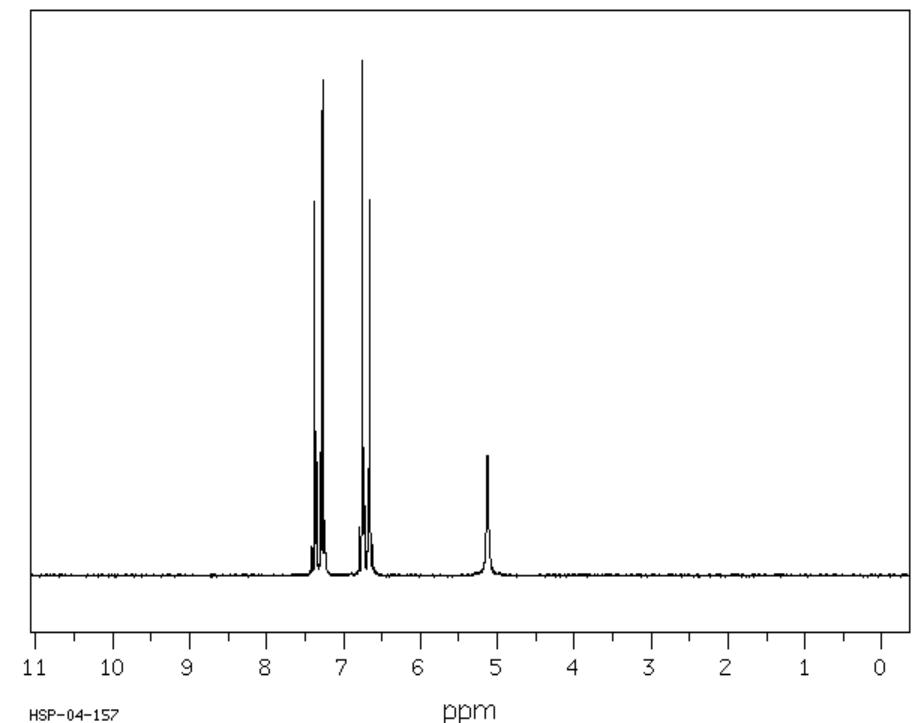
Increasing  $B_0$  the spectrum gets closer to first order.  
Why?

## Complicated spin systems: pl. AA' BB'

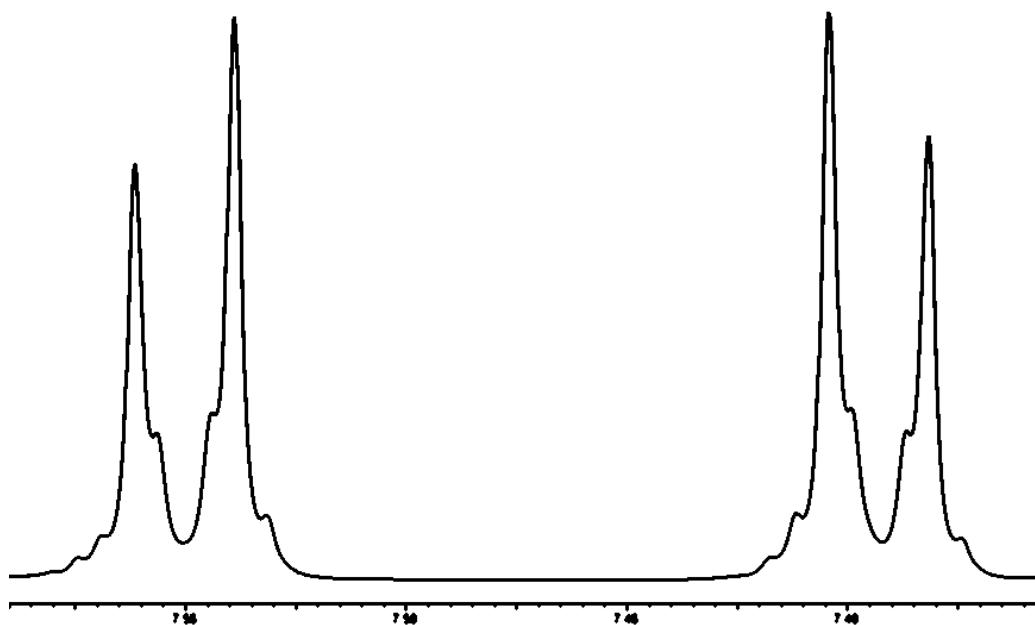
*p*- v. *o*-disubstituted benzene (chemically equivalent but magnetically not equivalent)

*p*-bromophenol

1,2-dibromobenzene



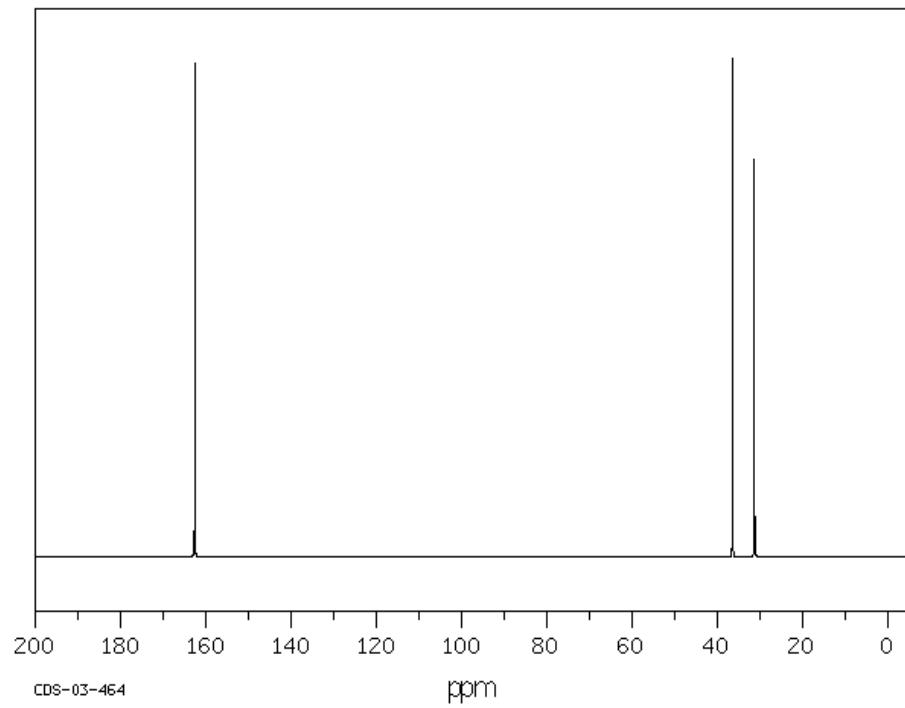
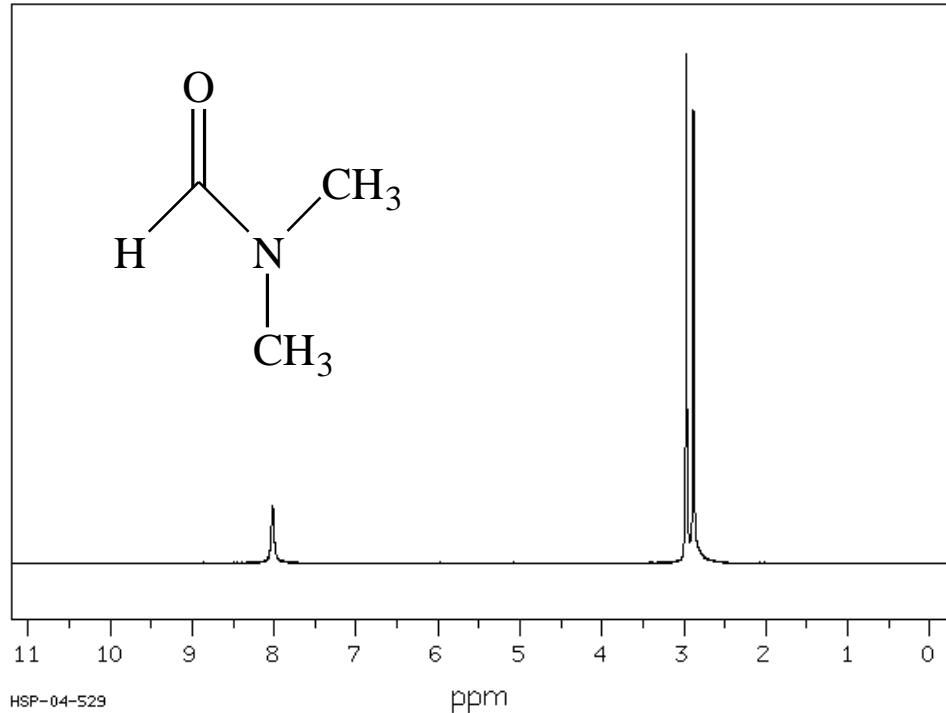
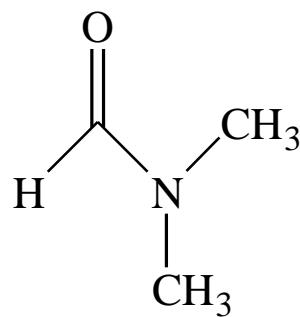
THF behaves similarly!



# Magnetic anisotropy

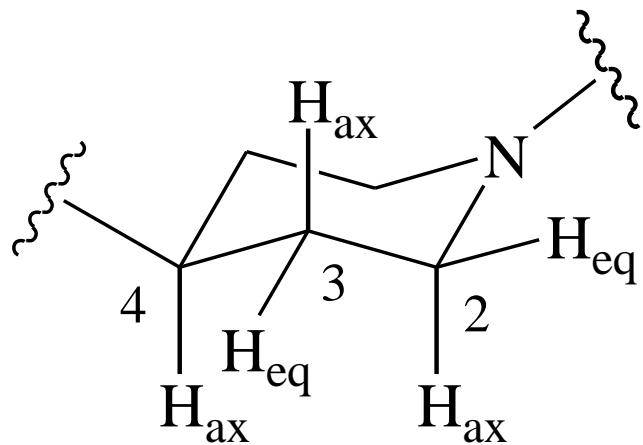
Neighbouring groups have effect on chemical shift through space as well (not just through chemical bond)

DMF



# Magnetic anisotropy

## Ring positions

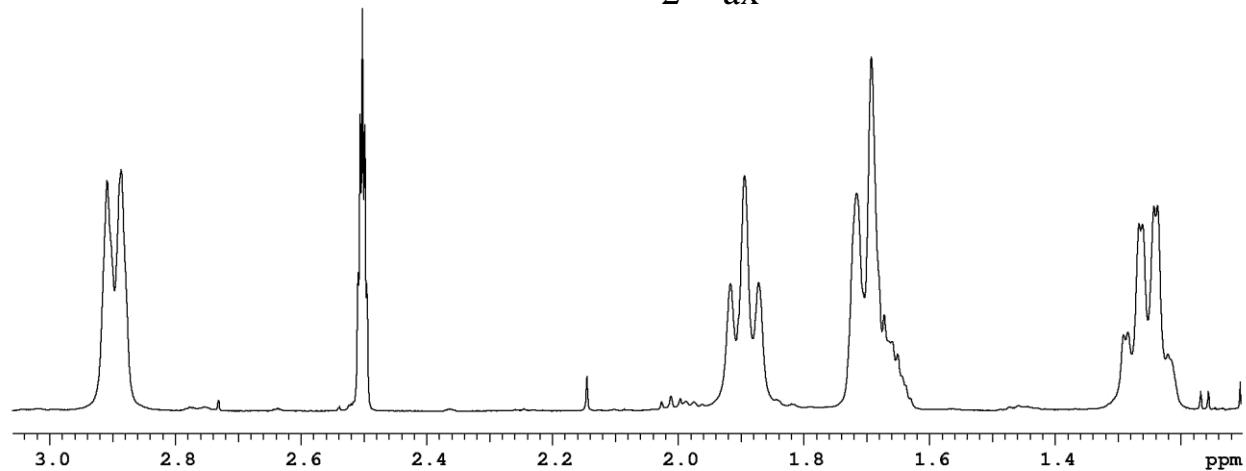


$_3\text{H}_{\text{eq}} + 4\text{H}$

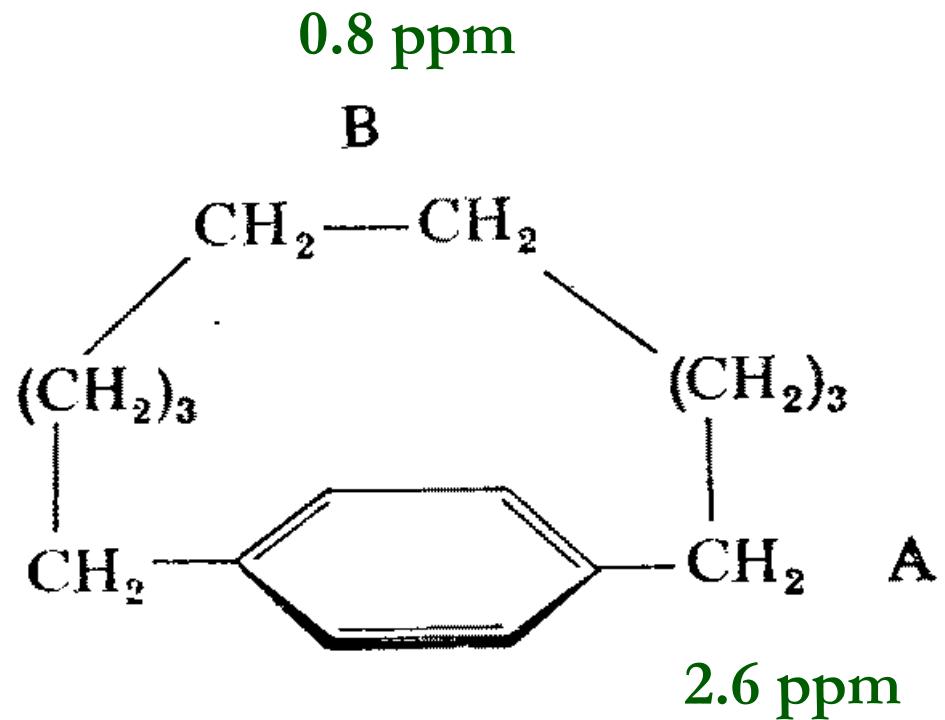
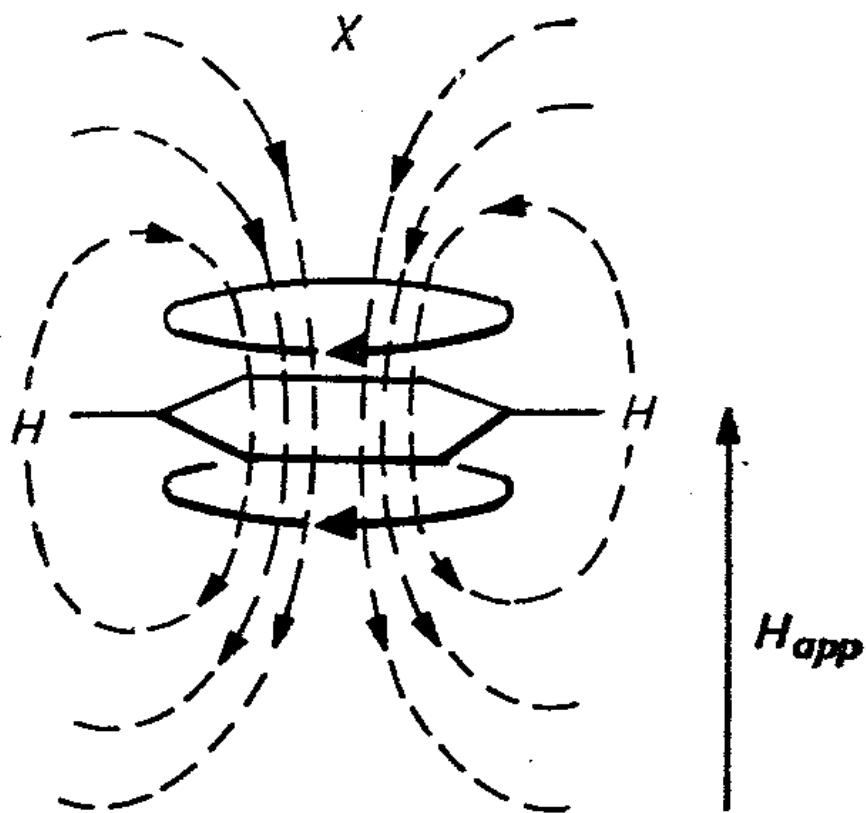
$2\text{H}_{\text{eq}}$

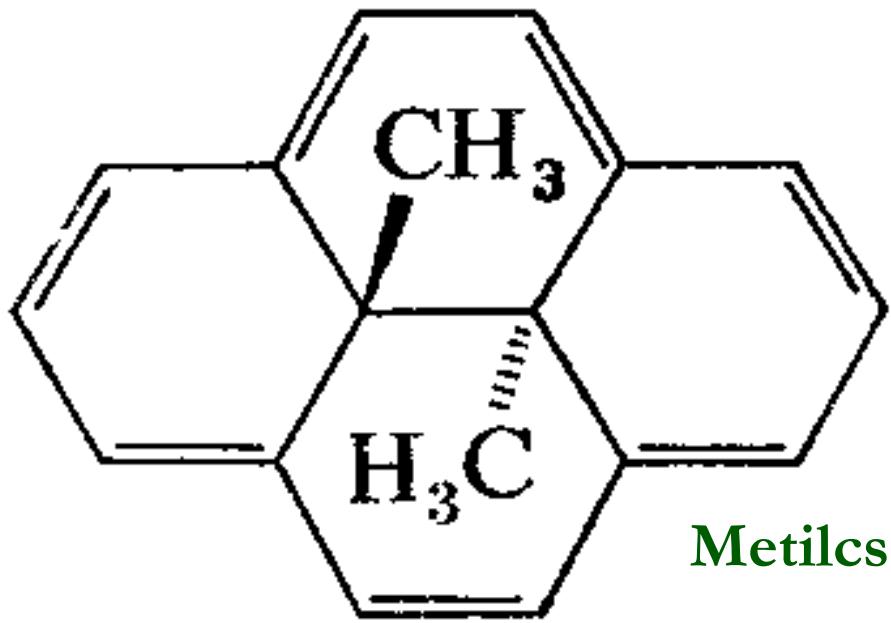
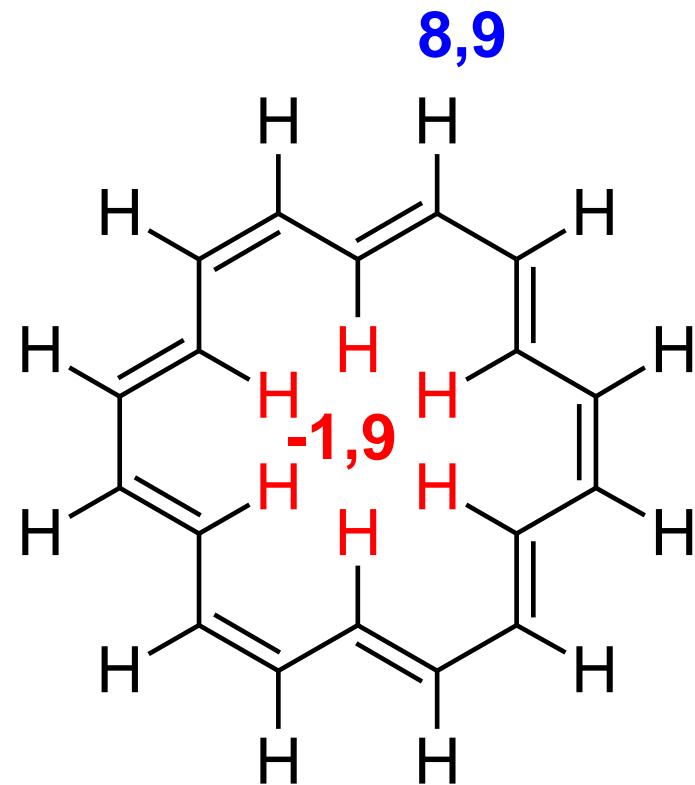
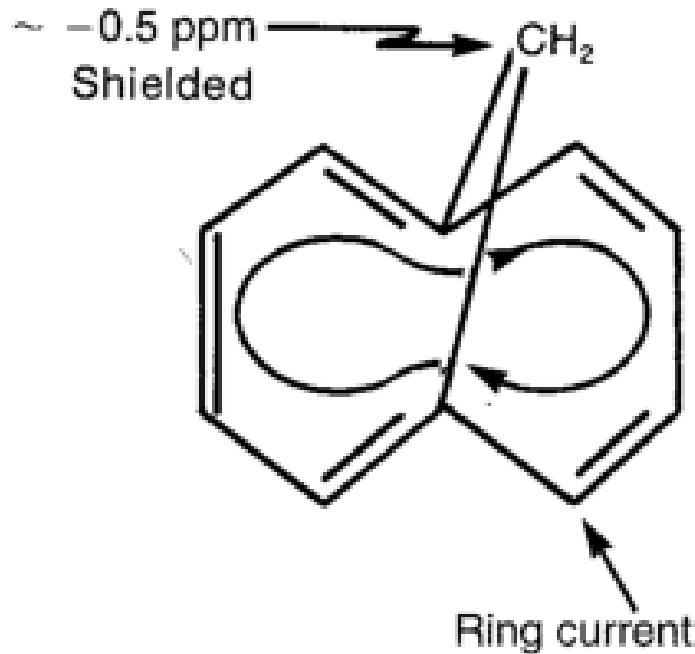
$2\text{H}_{\text{ax}}$

$3\text{H}_{\text{ax}}$

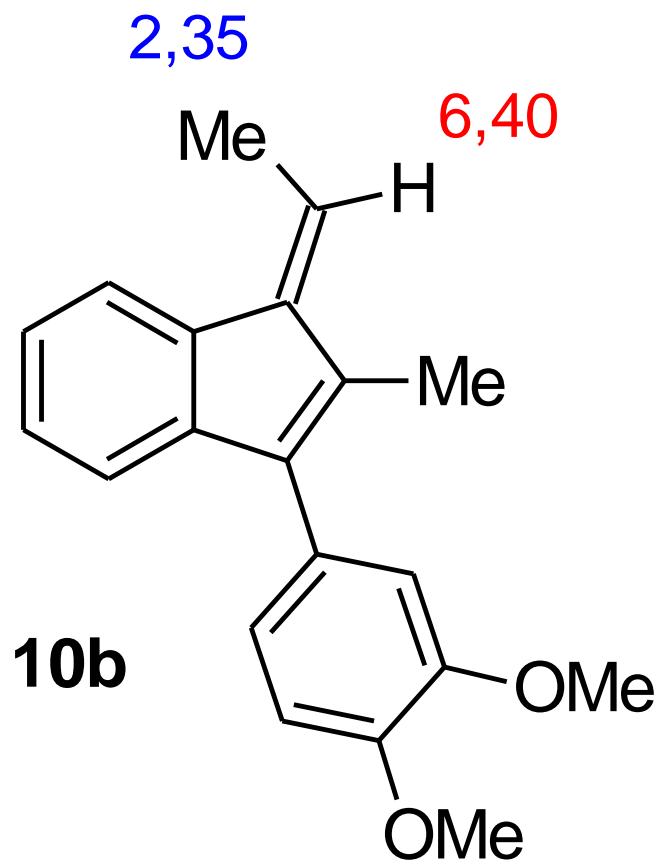
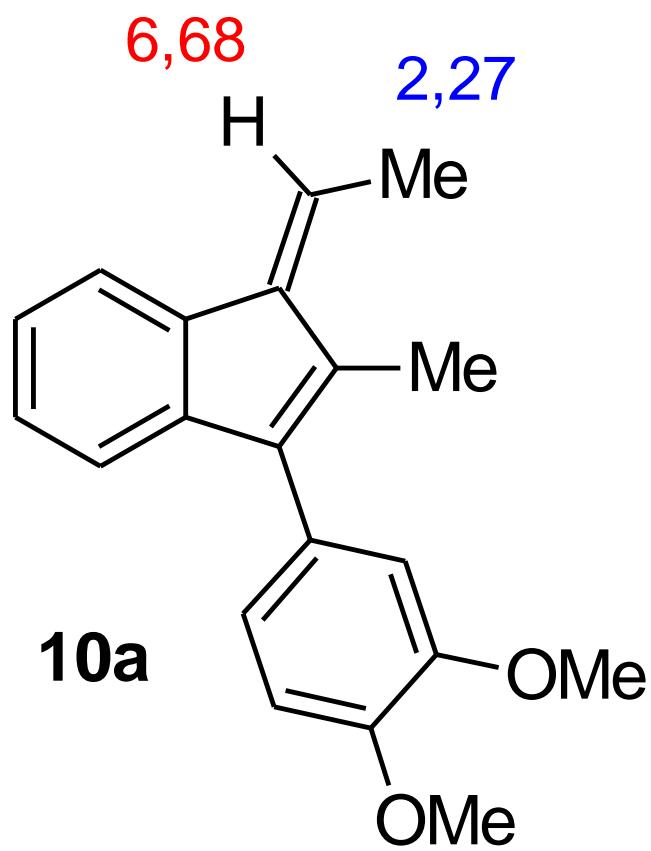


# Aromatic ring currents

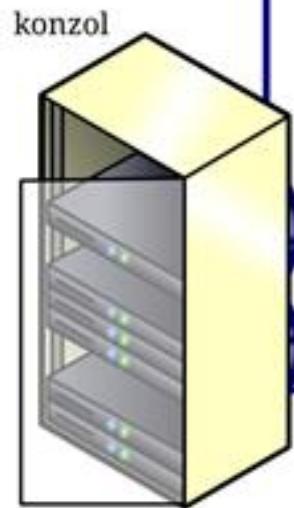
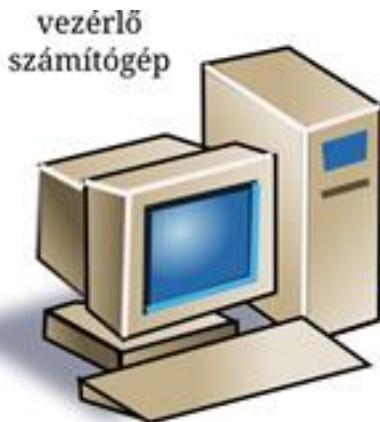




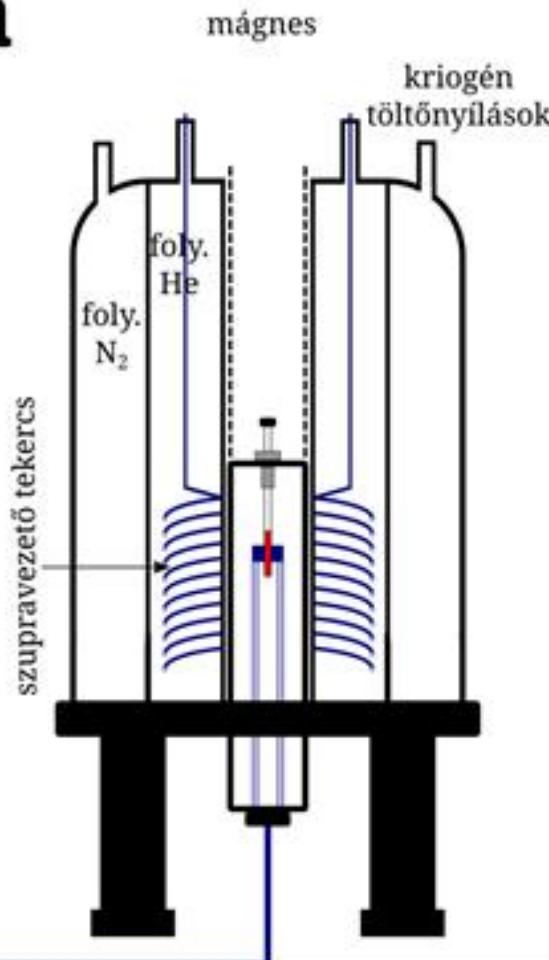
Metilcsoport: -4.2 ppm



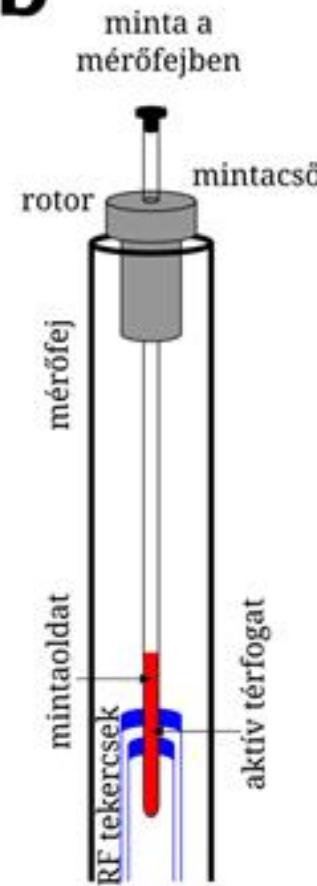
# NMR spectrometer



**a**



**b**



# How to measure?

1. Sample preparation
  1. Solvent
  2. O<sub>2</sub> exemption
  3. Filtering
2. Measurement
  1. Setting the instrument: tune, lock, shim
  2. Parameters
3. Data management
  1. FT
  2. Phase correction
  3. Baseline correction (BC)
  4. Referencing (TMS, solvent)
  5. Integral (<sup>1</sup>H)



(...administrating...)

$$\nu = \frac{\gamma}{2\pi} \cdot B$$

Is B constant (not taking chemical shift in account)?

**NO!**

The coil defines **H** (magnetic field)

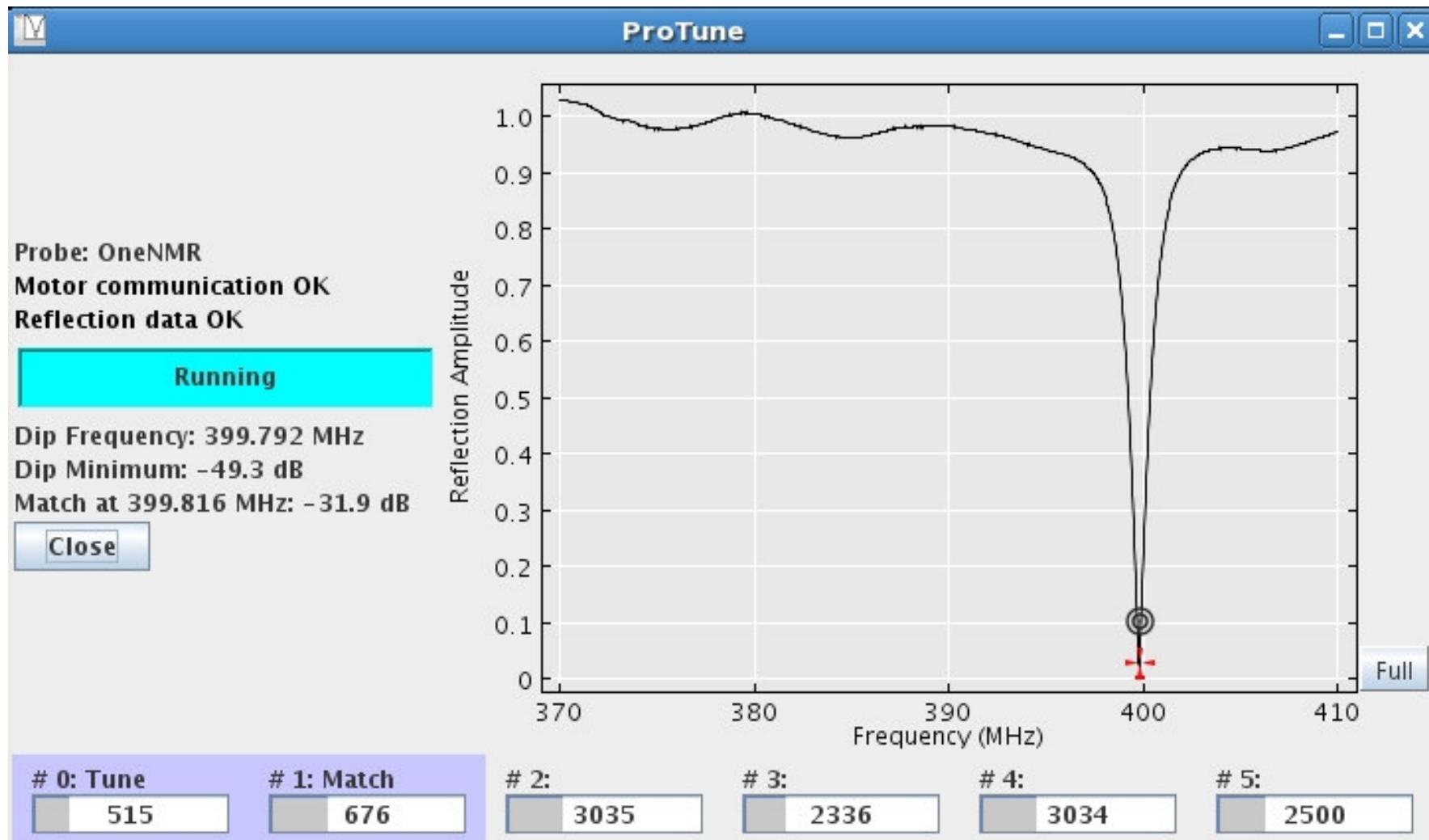
The magnetic induction (**B**) depends on the material in the magnetic field!!!

$$B = mH \quad \text{Magnetic permeability}$$

Does the material changes in the instrument???

# YES!!!

Tuning: RF coils parameters set depending on tube and solvent

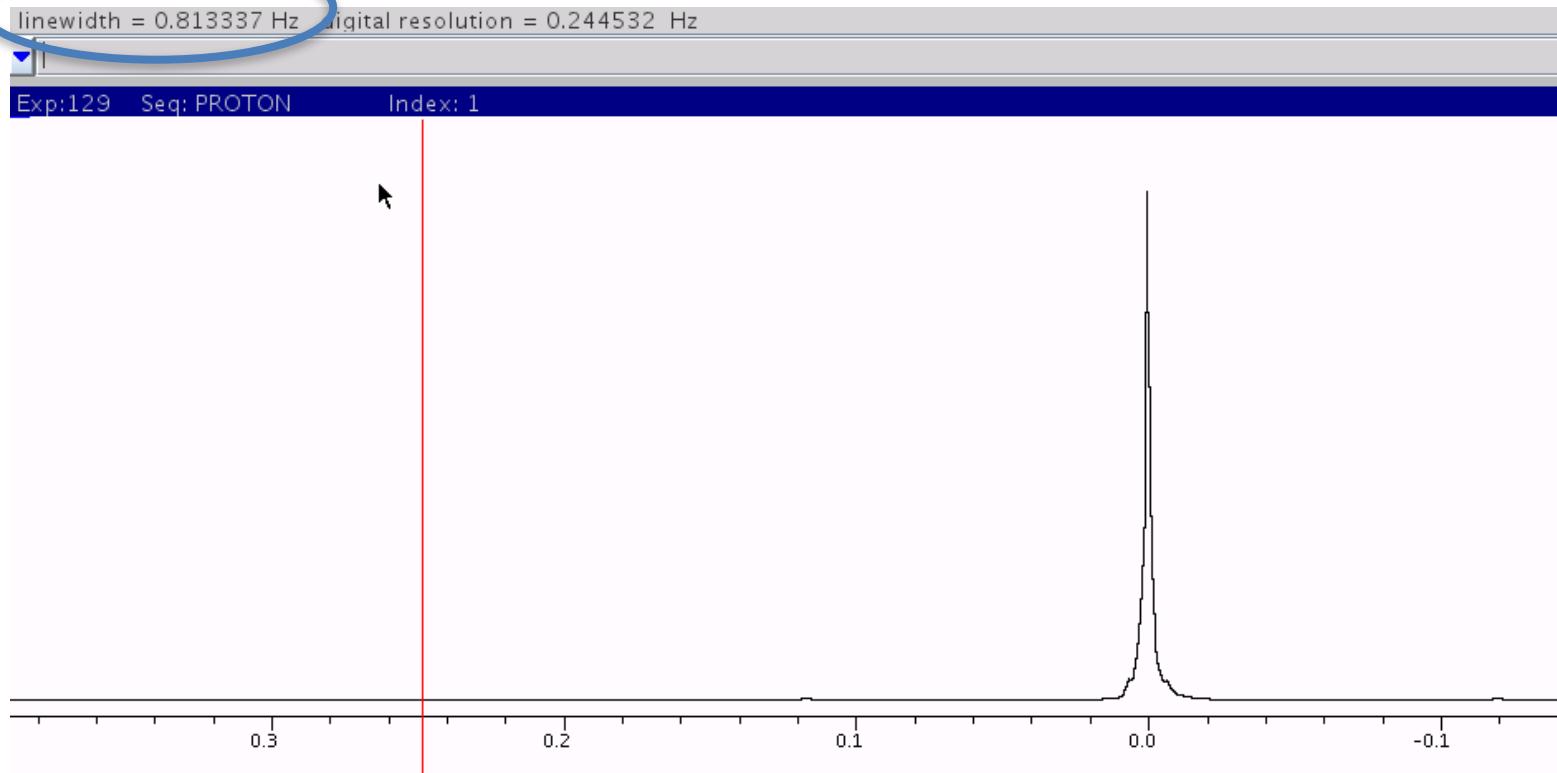


Lock: Solvent D signal to maintain B constant

# Shimming: magnetic field homogeneity

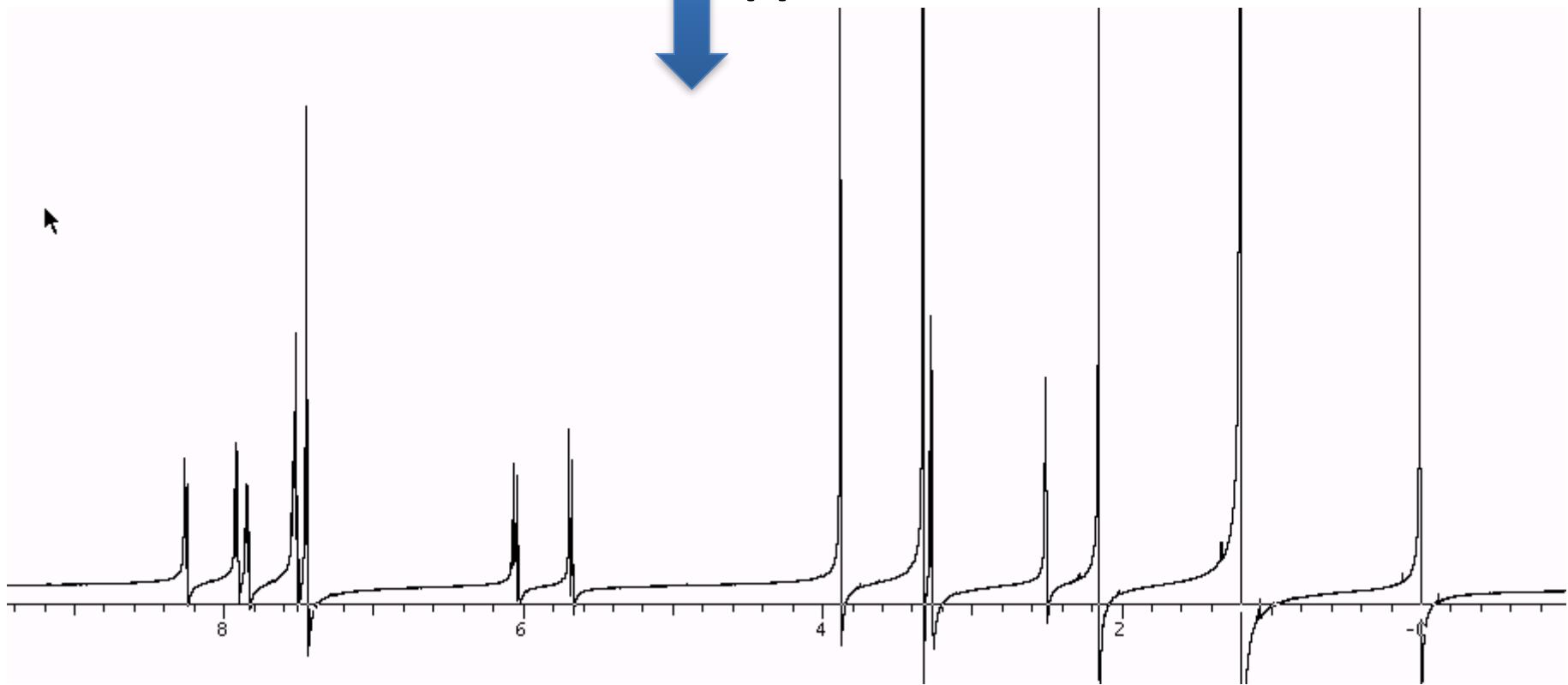


Control: singulet signal (eg. TMS) width

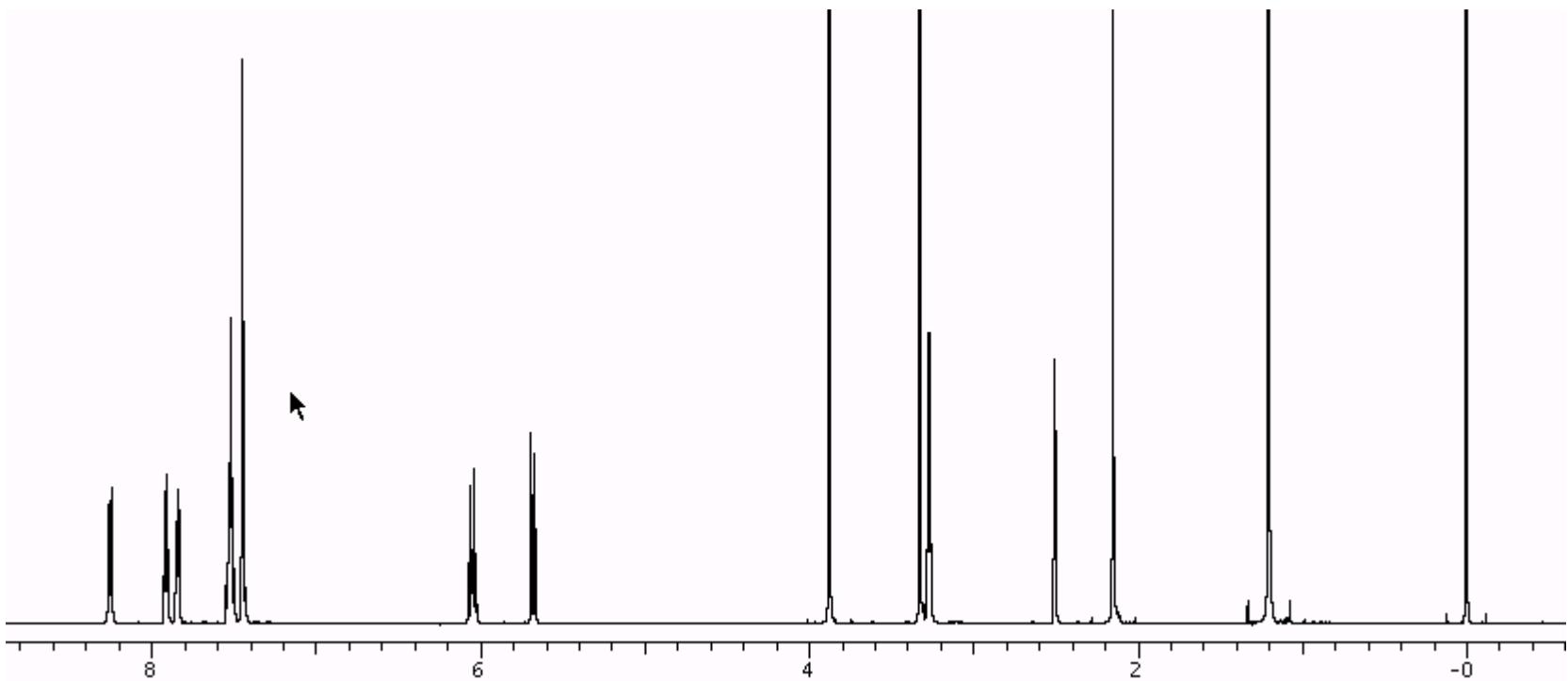


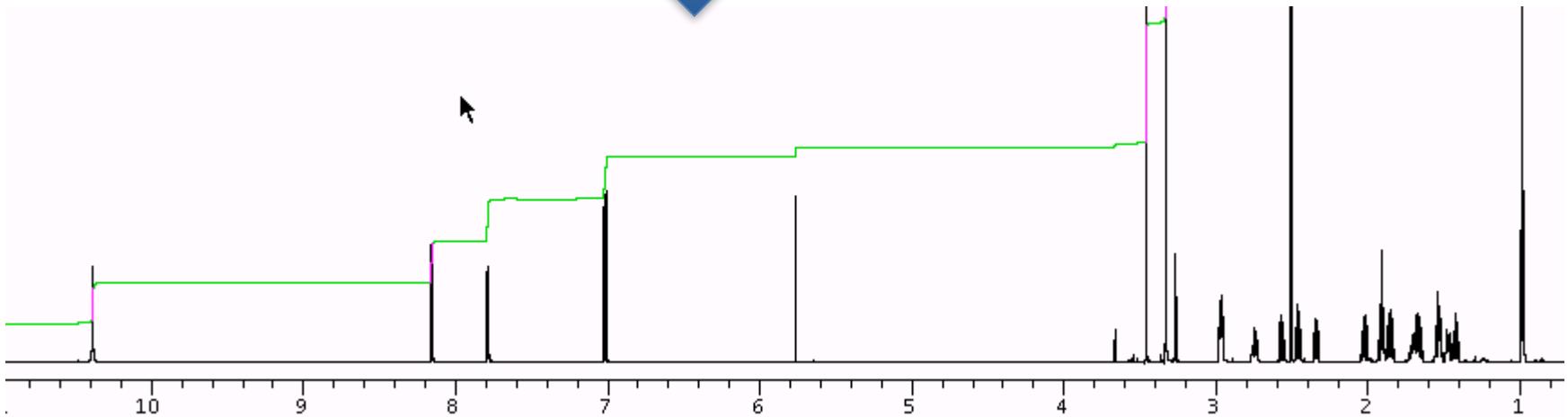
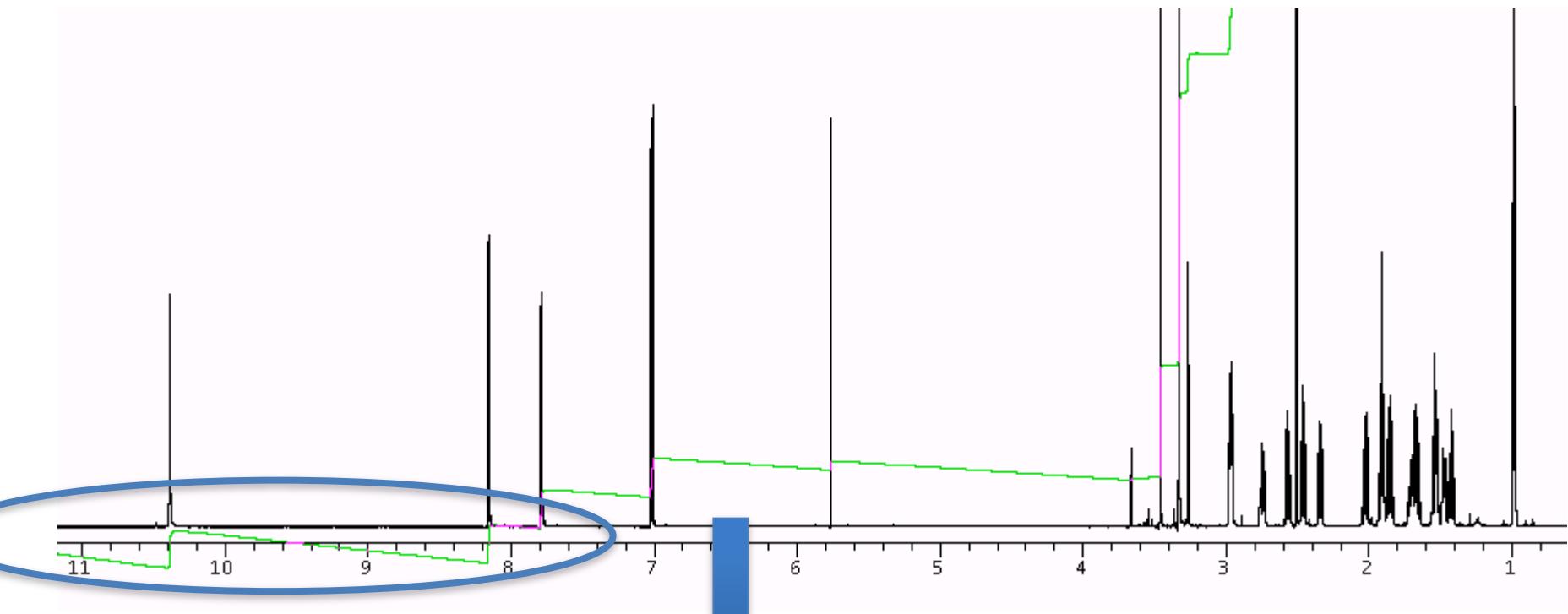


FT



## Phase correction





# Integration

